

## PATENT COOPERATION TREATY

## PCT

## INTERNATIONAL PRELIMINARY EXAMINATION REPORT

(PCT Article 36 and Rule 70)

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Applicant's or agent's file reference s021gm-1c21		FOR FURTHER ACTION	See Notification of Transmittal of International Preliminary Examination Report (Form PCT/IPEA/416)
International application No. <b>PCT/KR2002/002305</b>	International filing date (day/month/year) <b>07 DECEMBER 2002 (07.12.2002)</b>	Priority date (day/month/year) 06 DECEMBER 2002 (06.12.2002)	
International Patent Classification (IPC) or national classification and IPC  <b>IPC7 H01J 17/49</b>			
Applicant  <b>LG MICRON LTD. et al</b>			

1. This international preliminary examination report has been prepared by this International Preliminary Examining Authority and is transmitted to the applicant according to Article 36.



2. This REPORT consists of a total of 3 sheets, including this cover sheet.

☒ This report is also accompanied by ANNEXES, i.e., sheets of the description, claims and/or drawings which have been amended and are the basis for this report and/or sheets containing rectifications made before this Authority (see Rule 70.16 and Section 607 of the Administrative Instructions under the PCT).

These annexes consist of a total of 44 sheets.

3. This report contains indications relating to the following items:

- I ☒ Basis of the report
- II ☐ Priority
- III ☐ Non-establishment of opinion with regard to novelty, inventive step and industrial applicability
- IV ☐ Lack of unity of invention
- V ☒ Reasoned statement under Article 35(2) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement
- VI ☐ Certain documents cited
- VII ☐ Certain defects in the international application
- VIII ☐ Certain observations on the international application

Date of submission of the demand  <b>02 JULY 2004 (02.07.2004)</b>	Date of completion of this report  14 MARCH 2005 (14.03.2005)
Name and mailing address of the IPEA/KR  Korean Intellectual Property Office 920 Dunsan-dong, Seo-gu, Daejeon 302-701, Republic of Korea Facsimile No. 82-42-472-7140	Authorized officer  AHN, JOON JYUNG Telephone No. 82-42-481-5984 

# INTERNATIONAL PRELIMINARY EXAMINATION REPORT

International application No.

PCT/KR2002/002305

## I. Basis of the report

### 1. With regard to the elements of the international application:\*

- ☐ the international application as originally filed
- ☒ the description:  
 pages \_\_\_\_\_, as originally filed  
 pages \_\_\_\_\_, filed with the demand  
 pages 1-40 \_\_\_\_\_, filed with the letter of 28-01-2005
- ☒ the claims:  
 pages \_\_\_\_\_, as originally filed  
 pages \_\_\_\_\_, as amended (together with any statment) under Article 19  
 pages \_\_\_\_\_, filed with the demand  
 pages 41-44 \_\_\_\_\_, filed with the letter of 28-01-2005
- ☐ the drawings:  
 pages \_\_\_\_\_, as originally filed  
 pages \_\_\_\_\_, filed with the demand  
 pages \_\_\_\_\_, filed with the letter of \_\_\_\_\_
- ☐ the sequence listing part of the description:  
 pages \_\_\_\_\_, as originally filed  
 pages \_\_\_\_\_, filed with the demand  
 pages \_\_\_\_\_, filed with the letter of \_\_\_\_\_

### 2. With regard to the language, all the elements marked above were available or furnished to this Authority in the language in which the international application was filed, unless otherwise indicated under this item.

These elements were available or furnished to this Authority in the following language English which is

- ☐ the language of a translation furnished for the purposes of international search (under Rule 23.1(b)).
- ☒ the language of publication of the international application (under Rule 48.3(b)).
- ☐ the language of the translation furnished for the purposes of international preliminary examination (under Rules 55.2 and/or 55.3).

### 3. With regard to any nucleotide and/or amino acid sequence disclosed in the international application, the international preliminary examination was carried out on the basis of the sequence listing:

- ☐ contained in the international application in written form.
- ☐ filed together with the international application in computer readable form.
- ☐ furnished subsequently to this Authority in written form.
- ☐ furnished subsequently to this Authority in computer readable form
- ☐ The statement that the subsequently furnished written sequence listing does not go beyond the disclosure in the international application as filed has been furnished.
- ☐ The statement that the information recorded in computer readable form is identical to the written sequence listing has been furnished.

### 4. ☐ The amendments have resulted in the cancellation of:

- ☐ the description, pages \_\_\_\_\_
- ☐ the claims, Nos. \_\_\_\_\_
- ☐ the drawings, sheets \_\_\_\_\_

### 5. ☐ This report has been established as if (some of) the amendments had not been made, since they have been considered to go beyond the disclosure as filed, as indicated in the Supplemental Box (Rule 70.2(c)).\*\*

\* Replacement sheets which have been furnished to the receiving Office in response to an invitation under Article 14 are referred to in this opinion as "originally filed," and are not annexed to this report since they do not contain amendments (Rules 70.16 and 70.17).

\*\* Any replacement sheet containing such amendments must be referred to under item I and annexed to this report.

## INTERNATIONAL PRELIMINARY EXAMINATION

International application No.

PCT/KR2002/002305

**V. Reasoned statement under Article 35(2) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement****1. Statement**

Novelty (N)	Claims	1	YES
	Claims	None	NO
Inventive step (IS)	Claims	1	YES
	Claims	None	NO
Industrial applicability (IA)	Claims	1	YES
	Claims	None	NO

**2. Citations and explanations (Rule 70.7)****1. Reference is made to the following documents from the International Search Report(ISR):**

D1: WO99/24999A1

D2: JP2000-169764A (JP12-169764A)

2. The present application relates to a rear plate of a plasma display panel comprising, a glass substrate, electrodes, a dielectric layer, barrier ribs formed through etching, and phosphor layers with each part having a specific composition, electrical properties, a thermal expansive coefficient, a etching rate, a average particle diameter of raw materials and a color temperature.

3. D1 discloses a plasma display panel comprising a phosphor material of BaSrMgAlO:Eu and D2 discloses a plasma display panel comprising a dielectric layer made from the glass paste of a glass powder and a binder resin.

4. In view of the prior art, the subject matter of claim 1 appears to be novel in the sense of Article 33(2) PCT.

5. None of the documents cited in the ISR discloses a specific composition, electrical properties, a thermal expansive coefficient, a etching rate and a average particle diameter of raw materials, a color temperature of whole individual parts composing a rear plate of a plasma display panel. Since these teaching are not rendered obvious from the prior art, the subject matter of claim 1 appears to involve an inventive step in the sense of article 33(3) PCT.

6. The subject matter of claim 1 appears to be industrially applicable in the sense of Article 33(4) PCT.

REAR PLATE FOR PLASMA DISPLAY PANEL

**Technical Field**

5 The present invention relates to a rear plate of a plasma display panel.

**Background Art**

10 As generally known in the art, a plasma display panel (PDP) is a display device having a front glass substrate and a rear glass substrate between which a discharge space is formed, so that plasma discharge may generate in the discharge space, thereby causing phosphors in the discharge space to be excited and emit light, so as to display a screen.

15 PDPs may be classified into direct current plasma display panels (DC PDPs) and alternating current plasma display panels (AC PDPs), from among which the AC PDPs are the mainstream. U.S. Patent No. 5,446,344 discloses a three-electrode surface-discharge alternating-current plasma display panel which is one of the representative AC PDPs.

20 A PDP includes a front plate and a rear plate assembled in parallel with each other. The front plate includes a front glass substrate, transparent electrodes formed on a lower surface of the front glass substrate, each of the transparent electrodes including a scan electrode and a sustain electrode, bus electrodes formed on lower surfaces of the transparent electrodes so as to reduce resistance of the

transparent electrodes, a dielectric layer covering the transparent electrodes and the bus electrodes, and a magnesium oxide layer formed on a lower surface of the dielectric layer so as to prevent sputtering of the dielectric layer and facilitate discharge of secondary electrons. Further, the rear plate includes a rear glass substrate, address electrodes, a dielectric layer, barrier ribs for forming discharge compartments between the front and rear plates, and phosphorous layers.

In general, a rear plate of a PDP as described above is manufactured by sand blasting similar to a method of forming a thick film of a substrate of a PDP, which is disclosed by Japanese Patent Laid-Open No. P5-128966.

In the conventional rear plate manufactured using the sand blasting as described above, barrier ribs are preliminary formed by sand blasting and are then completed through firing. As a result, in the course of the firing process, the barrier ribs may be distorted and deformed. Therefore, it is difficult to exactly locate each electrode on a central position between two barrier ribs, which is a desired position for each electrode.

#### Disclosure of the Invention

Therefore, the present invention has been made in view of the above-mentioned problems, and it is an object of the present invention to provide a rear plate

of a plasma display panel, in which, not only can each electrode exactly be located on a central portion between barrier ribs, but also can various characteristics of a PDP be improved.

5       According to an aspect of the present invention, there is provided a rear plate of a plasma display panel, the rear plate comprising: a glass substrate; electrodes formed on an upper surface of the glass substrate; a dielectric layer formed on upper surfaces  
10 of the electrode and the upper surface of the glass substrate; barrier ribs formed in a shape of a pattern through etching on an upper surface of the dielectric layer; and phosphorous layers formed on side surfaces and bottom surfaces of the barrier ribs and including  
15 red, green, and blue phosphorous layers, which emit red, green, and blue light according to an electric signal, respectively, wherein: the electrodes are made from mixture of a conductive metal powder of 51 to 99.5 wt% and a first glass powder of 0.5 to 49 wt%, the  
20 conductive metal powder being at least one metal powder selected from metal powders of Au, Ag, Pt, Pd, Ni, and Cu; the conductive metal powder having an average particle diameter of 0.1 to 7  $\mu\text{m}$ , the first glass powder having an average particle diameter of 0.5 to 10  
25  $\mu\text{m}$  and a specific resistance of  $1.0 \times 10^{-6}$  to  $5.0 \times 10^{-6}$   $\Omega\text{cm}$ ; the dielectric layer is made from mixture of a first filler and at least one glass powder selected from among a second glass powder and a third glass powder, the second glass powder including PbO of 30 to

80 wt%, ZnO of 0 to 20 wt%, SiO<sub>2</sub> of 0 to 20 wt%, B<sub>2</sub>O<sub>3</sub> of 5 to 40 wt%, Al<sub>2</sub>O<sub>3</sub> of 0 to 12 wt%, Na<sub>2</sub>O+K<sub>2</sub>O+Li<sub>2</sub>O of 0 to 5 wt%, and BaO+CaO+MgO+SrO of 0 to 5 wt%; the third glass powder including Bi<sub>2</sub>O<sub>3</sub> of 36 to 84 wt%, B<sub>2</sub>O<sub>3</sub> of 5 to 28  
5 wt%, PbO of 0 to 46 wt%, ZnO of 0 to 30 wt%, Al<sub>2</sub>O<sub>3</sub> of 0 to 13 wt%, SiO<sub>2</sub> of 0 to 10 wt%, Na<sub>2</sub>O+K<sub>2</sub>O+Li<sub>2</sub>O of 0 to 5 wt%, and BaO+CaO+MgO+SrO of 0 to 3 wt%, each of the second and third glass powders having an average particle diameter of 0.5 to 10 µm, a softening  
10 temperature of 390 to 550 °C, a thermal expansive coefficient of  $63 \times 10^{-7}$  to  $83 \times 10^{-7}/^{\circ}\text{C}$ , a dielectric constant of 11 to 26, and an etching rate of 0.1 to 1.0 µm/min, the first filler having an average particle diameter of 0.5 to 10 µm and including at least one  
15 oxide selected from the group consisting of TiO<sub>2</sub>, ZrO<sub>2</sub>, ZnO, Al<sub>2</sub>O<sub>3</sub>, BN, SiO<sub>2</sub>, and MgO, which are white oxides, a ratio of volume of the first filler with respect to volume of the glass powder in the dielectric layer being 0.05 to 0.30, thereby the dielectric layer having  
20 a dielectric constant of 11 to 26, a reflectance of 50 to 80%, an etching rate of 0.1 to 1.0 µm/min, and a porosity of 5, when the dielectric layer has been baked for 10 to 60 minutes at 450 to 600 °C; the barrier ribs are made from mixture of at least one glass powder  
25 selected from the group consisting of the fourth, fifth, and sixth glass powders and at least one filler selected from the group consisting of a second filler and a third filler, the fourth glass powder including ZnO of 0 to 48 wt%, SiO<sub>2</sub> of 0 to 21 wt%, B<sub>2</sub>O<sub>3</sub> of 25 to 56

wt%,  $\text{Al}_2\text{O}_3$  of 0 to 12 wt%,  $\text{Na}_2\text{O}+\text{K}_2\text{O}+\text{Li}_2\text{O}$  of 0 to 38 wt%,  
and  $\text{BaO}+\text{CaO}+\text{MgO}+\text{SrO}$  of 0 to 15 wt%, the fifth glass  
powder including  $\text{PbO}$  of 25 to 65 wt%,  $\text{ZnO}$  of 0 to 35  
wt%,  $\text{SiO}_2$  of 0 to 26 wt%,  $\text{B}_2\text{O}_3$  of 5 to 30 wt%,  $\text{Al}_2\text{O}_3+\text{SnO}_2$   
5 of 0 to 13 wt%,  $\text{Na}_2\text{O}+\text{K}_2\text{O}+\text{Li}_2\text{O}$  of 0 to 19 wt%,  $\text{BaO}$  of 0 to  
26 wt%, and  $\text{CaO}+\text{MgO}+\text{SrO}$  of 0 to 13 wt%, the sixth glass  
powder including  $\text{PbO}$  of 35 to 55 wt%,  $\text{B}_2\text{O}_3$  of 18 to 25  
wt%,  $\text{ZnO}$  of 0 to 35 wt%,  $\text{BaO}$  of 0 to 16 wt%,  
 $\text{SiO}_2+\text{Al}_2\text{O}_3+\text{SnO}_2$  of 0 to 9 wt%,  $\text{CoO}+\text{CuO}+\text{MnO}_2+\text{Fe}_2\text{O}_3$  of 0 to  
10 15 wt%,  $\text{Na}_2\text{O}+\text{K}_2\text{O}+\text{Li}_2\text{O}$  of 0 to 19 wt%, and  $\text{CaO}+\text{MgO}+\text{SrO}$  of  
0 to 13 wt%, each of the fourth, fifth, and sixth glass  
powders has an average particle diameter of 0.5 to 10  
 $\mu\text{m}$ , a softening temperature of 390 to 630  $^{\circ}\text{C}$ , a thermal  
expansive coefficient of  $63 \times 10^{-7}$  to  $83 \times 10^{-7}/^{\circ}\text{C}$ , a  
15 dielectric constant of 5 to 20, and an etching rate of  
2.0 to 50.0  $\mu\text{m}/\text{min}$ , the second filler including at  
least two oxides selected from the group consisting of  
 $\text{NiO}$ ,  $\text{Fe}_2\text{O}_3$ ,  $\text{CrO}$ ,  $\text{MnO}_2$ ,  $\text{CuO}$ ,  $\text{Al}_2\text{O}_3$ , and  $\text{SiO}_2$ , which have  
dark colors, the third filler including at least one  
20 oxide selected from the group consisting of  $\text{TiO}_2$ ,  $\text{ZrO}_2$ ,  
 $\text{ZnO}$ ,  $\text{Al}_2\text{O}_3$ ,  $\text{BN}$ ,  $\text{SiO}_2$ , and  $\text{MgO}$ , which have white colors,  
each of the second and third fillers having an average  
particle diameter of 0.1 to 10  $\mu\text{m}$ , a ratio of the  
volume of the filler with respect to the volume of the  
25 glass powder for the barrier ribs being 0.05 to 0.67,  
thereby the barrier ribs having a dielectric constant  
of 5 to 16 and an etching rate of 2 to 50  $\mu\text{m}/\text{min}$  and  
enabling the glass substrate having the barrier ribs to  
have a bending of at most 0.3 mm, when the barrier ribs



has been baked for 10 to 60 minutes at 450 to 600 °C, the barrier ribs having a height difference of at most 1% when the barrier ribs has been baked at 510°C for one hour after being etched by acid-based etching solution, the barrier ribs having a destruction ratio of 50% when an iron rod, which weighs 500g and has an end portion shaped like a sphere having a radius of 3 mm, is dropped one hundred times vertically onto uppermost surfaces of the barrier ribs from 5 mm above the uppermost surfaces, each of the barrier ribs having at least one layer; and the red phosphorous layer includes at least two of YOx, GdOx, BOx, and EuOx, the green phosphorous layer includes at least one of ZnOx, SiOx, MnOx, YOx, BOx, TbOx, BaOx, and AlOx, and the blue phosphorous layer comprises at least two of BaOx, MgOx, AlOx, SrOx, MnOx, and EuOx, so that, in the phosphorous layers, color temperatures are maintained between 8,000K and 13,000K.

In the rear plate, barrier ribs are formed through etching after backing, and thus the completed barrier ribs are not deformed. Therefore, each electrode can be exactly located on a central portion between barrier ribs. When a PDP having front and rear plates attached to each other has been completed, optical characteristics of the PDP such as white brightness, color temperature, and contrast, and electric characteristics of the PDP such as voltage margin, power consumption, and electric efficiency, are improved, so that the reliability is improved.

**Brief Description of the Drawings**

The foregoing and other objects, features and advantages of the present invention will become more apparent from the following detailed description when taken in conjunction with the accompanying drawings in which:

FIG. 1 is a sectional view of a portion of a rear plate of a plasma display panel according to the present invention;

FIG. 2 is a graph showing an optical absorption ratio according to a ratio of volume of a filler with respect to a glass powder in a barrier rib of a rear plate according to an embodiment of the present invention; and

FIG. 3 is a graph showing an etching rate according to a ratio of volume of a filler with respect to a glass powder in a barrier rib of a rear plate according to an embodiment of the present invention.

**Best Mode for Carrying Out the Invention**

Hereinafter, a rear plate of a plasma display panel according to a preferred embodiment of the present invention will be described in detail with reference to the accompanying drawings.

As shown in FIG. 1, a rear plate 100 of a plasma display panel (hereinafter, referred to as "PDP") according to the present embodiment includes a glass substrate 110, electrodes 120 formed in a shape of a

pattern and spaced at a predetermined interval from each other on an upper surface of the glass substrate 110, a dielectric layer 130 formed on upper surfaces of the electrode 120 and the upper surface of the glass substrate 110, barrier ribs 140 formed on an upper surface of the dielectric layer 130 and spaced a predetermined interval from each other, and phosphorous layers 150 formed on side surfaces and bottom surfaces of the barrier ribs 140.

A method of manufacturing the barrier rib 140 according to the present embodiment will be briefly described hereinafter. First, a process of printing paste for barrier ribs on the entire upper surface of the dielectric layer 130 and then drying the paste is repeated several times, thereby forming a barrier rib layer. Then, the barrier rib layer is baked, a latent image is formed on the baked barrier rib layer by photolithography, and then the barrier rib layer is etched, so that the barrier ribs 140 are completed.

Since etching is utilized in forming the barrier ribs 140, the barrier rib layer must have a proper etching rate for etching solution and the electrodes 120 and the dielectric layer 130 must have resistance to the etching solution. In order to meet the requirements described above, each of the functional layers of the rear plate 100 according to the present embodiment has a specific composition, which will be described hereinafter.

The electrode 120 is made from a mixture of a

conductive metal powder and a first glass powder, which is a sintering agent for sintering the conductive metal powder at a low temperature. Preferably, the mixture includes a conductive metal powder of 51 to 99.5 wt% and a first glass powder of 0.5 to 49 wt%. When the mixture includes a conductive metal powder of less than 51 wt%, that is, when the mixture includes a first glass powder of more than 49 wt%, the resistance of the mixture is too high to lower the specific resistance of the electrode 120 under  $5.0 \times 10^{-6} \Omega\text{cm}$ , which will be described later in more detail. In contrast, when the mixture includes a conductive metal powder of more than 99.5 wt%, that is, when the mixture includes a first glass powder of less than 0.5 wt%, the proportion of the first glass powder is too small to enable sufficient sintering.

The conductive metal powder has an average particle diameter of 0.1 to 7  $\mu\text{m}$ . When the conductive metal powder has an average particle diameter of smaller than 0.1  $\mu\text{m}$ , the conductive metal powder has such a large specific surface area as to make it difficult to disperse the conductive metal powder. In contrast, when the conductive metal powder has an average particle diameter of at least 7  $\mu\text{m}$ , it is difficult to form an electrode having a thickness of at most 10  $\mu\text{m}$ , which is a proper thickness for the electrode 120. The first glass powder has an average particle diameter of 0.5 to 10  $\mu\text{m}$ . When the conductive metal powder has an average particle diameter of at

most 0.5  $\mu\text{m}$ , the conductive metal powder has such a large specific surface area as to make it difficult to disperse the conductive metal powder. In contrast, when the conductive metal powder has an average particle  
5 diameter of at least 10  $\mu\text{m}$ , it is difficult for the first glass powder to function as a binder for binding the conductive metal powder.

The electrode 120 made from a mixture of the conductive metal powder and the first glass powder has  
10 a specific resistance of  $1.0 \times 10^{-6}$  to  $5.0 \times 10^{-6} \Omega\text{cm}$ . When the electrode 120 has a specific resistance of less than  $1.0 \times 10^{-6} \Omega\text{cm}$ , the quantity of the conductive metal powder contained in the electrode 120 is excessive, thereby increasing the manufacturing cost of  
15 the electrode 120. In contrast, when the electrode 120 has a specific resistance of more than  $5.0 \times 10^{-6} \Omega\text{cm}$ , an address voltage, which is necessary in driving a PDP, becomes too high.

In order to possess the properties described  
20 above, the conductive metal powder includes at least one metal powder selected from the group consisting of powders of Au, Ag, Pt, Pd, Ni, and Cu, and the first glass powder includes a typical glass powder.

Hereinafter, the dielectric layer 130 formed on  
25 the electrodes 120 will be described. The dielectric layer 130 includes a first filler and at least one glass powder selected from among a second glass powder and a third glass powder.

It is preferred that each of the second and third

glass powders has an average particle diameter of 0.5 to 10  $\mu\text{m}$ . When each of the second and third glass powders has an average particle diameter of less than 0.5  $\mu\text{m}$ , they have a reduced workability. In contrast, when each of the second and third glass powders has an average particle diameter of more than 10  $\mu\text{m}$ , the dielectric layer 130 is not sufficiently compacted while being baked, so that the dielectric layer 130 may be porous.

It is also preferred that each of the second and third glass powders has a softening temperature of 390 to 550  $^{\circ}\text{C}$ . When their softening temperature is smaller than 390  $^{\circ}\text{C}$ , the dielectric layer 130 may flow in steps of firing the phosphorous layers and attaching the front plate and the rear plate of the PDP to each other after the barrier ribs 140 are formed, thereby deteriorating correctness in the measurements of the PDP. In contrast, when the softening temperature is larger than 550  $^{\circ}\text{C}$ , the firing temperature of the dielectric layer 130 increases to change the measurements of the glass substrate 110, thereby causing it difficult to control the measurements of the glass substrate 110.

Also, each of the second and third glass powders preferably has a thermal expansive coefficient of  $63 \times 10^{-7}$  to  $83 \times 10^{-7}/^{\circ}\text{C}$ . When the thermal expansive coefficient is smaller than  $63 \times 10^{-7}/^{\circ}\text{C}$ , the glass substrate 110 may be convexly bent. In contrast, when the thermal expansive coefficient is larger than  $83 \times$

10<sup>-7</sup>/°C, the glass substrate 110 may be concavely bent or the surface of the dielectric layer 130 may crack. However, even when each of the second and third glass powders preferably has a thermal expansive coefficient of 95 × 10<sup>-7</sup>/°C, the thermal expansive coefficient can be  
5 lowered to 83 × 10<sup>-7</sup>/°C by mixing a proper amount of the first filler with the second and third glass powder. Therefore, each of the second and third glass powders preferably may have a thermal expansive coefficient of  
10 63 × 10<sup>-7</sup> even up to 95 × 10<sup>-7</sup>/°C.

It is preferred that each of the second and third glass powders has a dielectric constant of 11 to 26. When the dielectric constant of the dielectric layer 130 is smaller than 11, it is difficult to transfer a  
15 signal of the electrode 120 to a discharge space defined by the barrier ribs 140. In contrast, when the dielectric constant of the dielectric layer 130 is larger than 26, the PDP has too slow a response speed when the PDP is driven. Meanwhile, when each of the  
20 second and third glass powders has a dielectric constant of at least 6, the dielectric constant of the dielectric layer 130 can be elevated up to 11 by means of the first filler. Therefore, it is also preferred that each of the second and third glass powders has a  
25 dielectric constant of 6 to 26.

Preferably, each of the second and third glass powders has an etching rate of 0.1 to 1.0 μm/min. When the etching rate is smaller than 0.1 μm/min, the firing temperature of the dielectric layer 130 may rise above

700 °C, thereby deforming the glass substrate 110. In contrast, when the etching rate is larger than 1.0  $\mu\text{m}/\text{min}$ , the powder has a reduced resistance to etching, so that even the dielectric layer 130 and the electrodes 120 may be etched when the barrier rib 140 is etched. When the electrode 120 is damaged by etching, the electric resistance of the electrode 120 increases.

It is preferred that a ratio of the volume of the first filler with respect to the volume of the glass powder in the dielectric layer is 0.05 to 0.30. When the volumetric ratio is less than 0.05, the dielectric layer 130 has a reflectance of at most 50%, thereby preventing the PDP from employing a dielectric layer having a reflectance of at least 50% which is necessary in order to enable the PDP to have an improved brightness. Further, in the case in which the volumetric ratio is more than 0.3, when the softening temperature of the glass powder is low, the dielectric constant is high and thus the response speed is slow. In contrast, when the softening temperature of the glass powder is high, the degree of firing of the dielectric layer 130 is deteriorated, so that it is difficult for the dielectric layer 130 to have a resistance to etching, and the dielectric layer 130 has a dielectric constant of at most 11.

In order to possess the properties as described above: the second glass powder comprises PbO of 30 to 80 wt%, ZnO of 0 to 20 wt%, SiO<sub>2</sub> of 0 to 20 wt%, B<sub>2</sub>O<sub>3</sub> of



5 to 40 wt%,  $\text{Al}_2\text{O}_3$  of 0 to 12 wt%,  $\text{Na}_2\text{O}+\text{K}_2\text{O}+\text{Li}_2\text{O}$  of 0 to 5 wt%, and  $\text{BaO}+\text{CaO}+\text{MgO}+\text{SrO}$  of 0 to 5 wt%; the third glass powder comprises  $\text{Bi}_2\text{O}_3$  of 36 to 84 wt%,  $\text{B}_2\text{O}_3$  of 5 to 28 wt%,  $\text{PbO}$  of 0 to 46 wt%,  $\text{ZnO}$  of 0 to 30 wt%,  $\text{Al}_2\text{O}_3$  of 0 to 13 wt%,  $\text{SiO}_2$  of 0 to 10 wt%,  $\text{Na}_2\text{O}+\text{K}_2\text{O}+\text{Li}_2\text{O}$  of 0 to 5 wt%, and  $\text{BaO}+\text{CaO}+\text{MgO}+\text{SrO}$  of 0 to 3 wt%; and the first filler has an average particle diameter of 10  $\mu\text{m}$  and includes at least one oxide selected from the group consisting of  $\text{TiO}_2$ ,  $\text{ZrO}_2$ ,  $\text{ZnO}$ ,  $\text{Al}_2\text{O}_3$ ,  $\text{BN}$ ,  $\text{SiO}_2$ , and  $\text{MgO}$ , which are white oxides.

When the second glass powder comprises  $\text{PbO}$  of less than 30 wt%, the softening temperature of the second glass powder becomes so high that the powder may lose fluidity and thus be insufficiently sintered. When the second glass powder comprises  $\text{PbO}$  of more than 80 wt%, the powder has such a high thermal expansive coefficient that the dielectric layer 130 may have a cracked surface or be bent. Further, when the second glass powder comprises  $\text{ZnO}$  of more than 20 wt% or  $\text{Na}_2\text{O}+\text{K}_2\text{O}+\text{Li}_2\text{O}$  of more than 5 wt%, the second glass powder may be crystallized. Further, when the second glass powder comprises either  $\text{SiO}_2$  of more than 20 wt%, or  $\text{Al}_2\text{O}_3$  of more than 12 wt%, or  $\text{BaO}+\text{CaO}+\text{MgO}+\text{SrO}$  of more than 5 wt%, the softening temperature of the second glass powder becomes so high that the powder may lose fluidity and thus be insufficiently sintered. Also, when the second glass powder comprises  $\text{B}_2\text{O}_3$  of less than 5 wt%, the softening temperature of the second glass powder becomes so high that the powder may lose

fluidity and thus be insufficiently sintered. In contrast, when the second glass powder comprises  $B_2O_3$  of more than 40 wt%, phase separation may be caused in the second glass powder.

5           When the third glass powder comprises  $Bi_2O_3$  of less than 36 wt%, the softening temperature of the third glass powder becomes so high that the powder may lose fluidity and thus be insufficiently sintered. When the third glass powder comprises  $Bi_2O_3$  of more than 84  
10 wt%, the softening temperature becomes too low. When the third glass powder comprises  $B_2O_3$  of less than 5 wt%, it is difficult to vitrify the dielectric layer 130. When the third glass powder comprises  $B_2O_3$  of more than 28 wt%, phase separation may be caused in the  
15 third glass powder. Further, when the third glass powder comprises PbO of more than 46 wt%, the powder has such a high thermal expansive coefficient that the dielectric layer 130 may have a cracked surface or be bent. Further, when the third glass powder comprises  
20 ZnO of more than 30 wt% or  $Na_2O+K_2O+Li_2O$  of more than 5 wt%, the third glass powder may be crystallized. Further, when the third glass powder comprises either  $SiO_2$  of more than 10 wt%, or  $Al_2O_3$  of more than 13 wt%, or  $BaO+CaO+MgO+SrO$  of more than 3 wt%, the softening  
25 temperature of the third glass powder becomes so high that the powder may lose fluidity and thus be insufficiently sintered.

The dielectric layer 130 having the ingredients as described above has a dielectric constant of 11 to 26,

a reflectance of 50 to 80%, and an etching rate of 0.1 to 1.0  $\mu\text{m}/\text{min}$ , when the dielectric layer 130 has been baked for 10 to 60 minutes at 450 to 600  $^{\circ}\text{C}$ . Further, the dielectric layer 130 has a porosity of 5%.

5           The reason because the dielectric layer 130 has a dielectric constant of 11 to 26 and an etching rate of 0.1 to 1.0  $\mu\text{m}/\text{min}$  is the same as that for the second and third glass powders. Further, when the dielectric layer 130 has a reflectance of less than 50%, the PDP  
10   has a deteriorated brightness. The dielectric layer 130 has a reflectance of at least 85% when the dielectric layer 130 contains a large quantity of the first filler or has been insufficiently baked. However, when the dielectric layer 130 has a reflectance of more than  
15   85%, it is difficult to obtain a desired etching rate. Therefore, the dielectric layer 130 may have a reflectance of 50 to 80%.

          Further, when the dielectric layer 130 has a porosity of more than 5%, the dielectric layer 130 may  
20   contain a relatively large bubble. Then, the dielectric layer 130 has a low withstand voltage, dielectric breakdown may be caused while the PDP is being driven.

          Next, a result of an experiment in which properties of the dielectric layer 130 having the  
25   composition as described above have been measured will be described.

          First, softening temperatures, etching rates, and dielectric constants of the second glass powder, which have been measured from various combinations of

ingredients of the second glass powder, will be described hereinafter.

Table 1

5 Properties of the second glass powder measured from various combinations of ingredients of the second glass powder:

2nd glass powder	Proportions of ingredients (wt%)							Properties		
	PbO	B <sub>2</sub> O <sub>3</sub>	ZnO	SiO <sub>2</sub>	Al <sub>2</sub> O <sub>3</sub>	BaO+ CaO+ MgO+ SrO	Na <sub>2</sub> O +K <sub>2</sub> O + Li <sub>2</sub> O	Soft- ening tempe- rature (°C)	Etch- ing rates (µm/min)	Dielec- -tric const- ants
Ex.1	32	39	6	16	6	0	1	527	0.87	6.2
Ex.2	53	8	19	3	10	2	5	485	0.63	8.5
Ex.3	61	30	6	1	2	0	0	467	0.95	15.4
Ex.4	44	26	10	12	5	1	2	511	0.23	12.7
Ex.5	75	10	3	4	4	4	0	432	0.98	20.3
Ex.6	53	6	10	13	12	2	0	535	0.12	13.2

10 As apparent from Table 1, examples 1 to 6 of the second glass powder have softening temperatures within a range between 390 and 550 °C, etching rates within a range between 0.1 and 1.0 µm/min, and dielectric constants within a range between 6 and 26.

15 Hereinafter, a method of measuring an etching rate will be described. First, the second glass powder according to one of examples 1 to 6 is applied on an entire upper surface of a substrate such as a glass

substrate, which is then baked. Then, acid-resistant tapes are attached to an upper surface of the baked second glass powder with intervals of 5 mm spaced from each other. Thereafter, the substrate is etched for ten minutes by acid-based etching solution, is washed for five minutes by ultrasonic waves and for one minute by flowing water, and is then dried. Thereafter, a depth by which the second glass powder has been etched is measured. Then, the etching rate is obtained by dividing the measured depth by etching time.

Next, softening temperatures, etching rates, and dielectric constants of the third glass powder, which have been measured from various combinations of ingredients of the second glass powder, will be described.

Table 2

Properties of the second glass powder measured from various combinations of ingredients of the third glass powder:

3rd Glass Powder	Proportions of ingredients (wt%)								Properties		
	Bi <sub>2</sub> O <sub>3</sub>	B <sub>2</sub> O <sub>3</sub>	PbO	ZnO	Al <sub>2</sub> O <sub>3</sub>	SiO <sub>2</sub>	Na <sub>2</sub> O+	BaO	Soft- ening temper- atures (°C)	Etching Rates (µm/min)	Diel- ectric Cons- tants
							K <sub>2</sub> O+	+	CaO		
							Li <sub>2</sub> O	+	MgO		
								+	SrO		
Ex.7	43	11	42	0	1	1	2	0	394	0.43	12.7

Ex.8	83	6	0	0	3	4	4	0	414	0.31	18.2
Ex.9	70	17	0	8	2	0	1	2	464	0.40	10.3
Ex.10	63	16	2	17	0	2	0	0	473	0.22	15.1
Ex.11	61	10	0	28	0	0	1	0	512	0.49	11.3
Ex.12	50	28	0	8	14	0	0	0	562	0.23	12.5

As apparent from Table 2, examples 7 to 12 of the third glass powder have softening temperatures within a range between 390 and 550 °C, etching rates within a range between 0.1 and 1.0 µm/min, and dielectric constants within a range between 6 and 26.

That is, as noted from Tables 1 and 2, the second and third glass powder have properties within desired ranges when they have ingredients combined according to proportions within the ranges described above.

Next, measured properties of the dielectric layers 130, which are produced by mixing the second glass powder according to example 3 and the third glass powder according to example 11 or 12 with the first filler and then firing them, will be described. In this case, TiO<sub>2</sub> is used as the first filler.

Table 3

Measured properties of a dielectric layer:

Class.	Glass powder					Dielectric layer				
Type of Dielectric	Type of Glass powder	Softening Temperature	Etch rate (µm/min)	Dielectric constant	Voluminal Ratio of Filler	Firing temperature (°C)	Dielectric constant	Refractivity	Porosity	Etch Rate (µm/min)

	er	(°C)			/glass powder				ity	
Ex.1	Ex.3	467	0.95	15.4	0.20	550	20.4	70	2	0.53
Ex.2	Ex.3	467	0.95	15.4	0.35	550	29	76	3	0.46
Ex.3	Ex.3	467	0.95	15.4	0.20	530	18.6	73	4	0.74
Ex.4	Ex.11	512	0.49	11.3	0.20	550	15.2	70	4	0.21
Ex.5	Ex.11	512	0.49	11.3	0.35	530	5.7	79	25	4.91
Ex.6	Ex.11	562	0.23	12.5	0.20	550	4.5	82	14	7.92

As shown in Table 3, the dielectric layer 130 according to example 1 is made by mixing the first filler with the second glass powder according to example 3 having a softening temperature within the range of 390 to 550 °C by a volumetric ratio of at most 0.3, which enables all properties of the dielectric layer 130 according to example 1 to be within the desired conditions described above. However, the dielectric layer 130 according to example 2 is made by the same mixture with a volumetric ratio of larger than 0.3, which causes the dielectric constant to be larger than 26, thereby making it difficult to use the dielectric layer 130 according example 2. The dielectric layer 130 according to example 3 is nearly the same as the dielectric layer 130 according to example 1, excepting that the firing temperature for firing the dielectric layer 130. As understood from the dielectric layers 130 according to example 1 and 3, properties of the dielectric layer 130 can be adjusted by adjusting the firing temperature of the dielectric

layer 130.

Further, the dielectric layer 130 according to example 4 is made by mixing the first filler with the third glass powder according to example 11 having a softening temperature within the range of 390 to 550 °C by a volumetric ratio of at most 0.3, and thus the dielectric layer 130 according to example 4 is usable. In contrast, the dielectric layer 130 according to example 5 is made by the same mixture with a volumetric ratio of larger than 0.3, which causes the dielectric constant of the dielectric layer 130 to be smaller than 6 and the etching rate to be larger than 1 µm/min, thereby making it difficult to use the dielectric layer 130 according to example 5.

Moreover, the dielectric layer 130 according to example 6 is made by mixing the first filler with the third glass powder according to example 12 having a softening temperature outside the range of 390 to 550 °C by a volumetric ratio of at most 0.3, which causes the dielectric constant of the dielectric layer 130 to be smaller than 6 and the etching rate to be larger than 1 µm/min, thereby making it difficult to use the dielectric layer 130 according to example 6.

Next, the barrier rib 140 formed on the upper surface of the dielectric layer 130 will be described.

The barrier rib 140 is made by mixing at least one glass powder selected from the group consisting of the fourth, fifth, and sixth glass powders with at least one filler selected from the group consisting of the



dark second filler and the white third filler. The barrier rib 140 includes one or more layers.

Each of the fourth, fifth, and sixth glass powders has an average particle diameter of 0.5 to 10  $\mu\text{m}$ . When  
5 the glass powder has an average particle diameter of smaller than 0.5  $\mu\text{m}$ , it is difficult to make paste for the barrier rib. In contrast, when the conductive metal powder has an average particle diameter of larger than 10  $\mu\text{m}$ , it is difficult to enable the barrier ribs to be  
10 sufficiently compact through firing after forming the barrier ribs.

Each of the fourth, fifth, and sixth glass powders has a softening temperature of 390 to 630  $^{\circ}\text{C}$ . In the case where the softening temperature is smaller than  
15 390  $^{\circ}\text{C}$ , when the phosphorous layers 150 are baked after the barrier ribs 140 are formed or after the front and rear plates are attached to each other, the barrier ribs 140 may be deformed so that the barrier ribs 140 may have much irregular heights and their upper  
20 portions may have much irregular widths. In contrast, in a case where the softening temperature is larger than 630  $^{\circ}\text{C}$ , the firing temperature of the barrier rib 140 increases to change the measurements of the glass substrate 110, thereby causing it difficult to control  
25 the measurements of the glass substrate 110.

Each of the fourth, fifth, and sixth glass powders preferably has a thermal expansive coefficient of  $63 \times 10^{-7}$  to  $83 \times 10^{-7}/^{\circ}\text{C}$ . When the thermal expansive coefficient is smaller than  $63 \times 10^{-7}/^{\circ}\text{C}$ , the glass

substrate 110 may be convexly bent. In contrast, when the thermal expansive coefficient is larger than  $83 \times 10^{-7}/^{\circ}\text{C}$ , the glass substrate 110 may be concavely bent or the surface of the glass substrate 110 may crack.

5 However, since the thermal expansive coefficient can be changed by adjusting the amount of the filler in the barrier rib 140, each of the fourth, fifth, and sixth glass powders preferably may have a thermal expansive coefficient of  $63 \times 10^{-7}$  even up to  $110 \times 10^{-7}/^{\circ}\text{C}$ .

10 It is preferred that each of the fourth, fifth, and sixth glass powders has a dielectric constant of 5 to 20. In the case where the dielectric constant is smaller than 5, a drive voltage characteristic is deteriorated when a manufactured PDP is driven. In  
15 contrast, in the case where the dielectric constant is larger than 20, crosstalk and erroneous discharge may happen when the manufactured PDP is driven.

Preferably, each of the fourth, fifth, and sixth glass powders has an etching rate of 2.0 to 50.0  
20  $\mu\text{m}/\text{min}$ . When the etching rate is smaller than 2.0  $\mu\text{m}/\text{min}$ , it takes too much time in forming the barrier ribs 140. Meanwhile, it is difficult to realize the etching rate of 50.0  $\mu\text{m}/\text{min}$  by the composition of the fourth, fifth, and sixth glass powder.

25 It is preferred that a ratio of the volume of the first filler with respect to the volume of the glass powder for the barrier ribs is 0.05 to 0.67, which will be described later.

In order to possess the properties as described

above: the fourth glass powder comprises ZnO of 0 to 48 wt%, SiO<sub>2</sub> of 0 to 21 wt%, B<sub>2</sub>O<sub>3</sub> of 25 to 56 wt%, Al<sub>2</sub>O<sub>3</sub> of 0 to 12 wt%, Na<sub>2</sub>O+K<sub>2</sub>O+Li<sub>2</sub>O of 0 to 38 wt%, and BaO+CaO+MgO+SrO of 0 to 15 wt%; the fifth glass powder  
5 comprises PbO of 25 to 65 wt%, ZnO of 0 to 35 wt%, SiO<sub>2</sub> of 0 to 26 wt%, B<sub>2</sub>O<sub>3</sub> of 5 to 30 wt%, Al<sub>2</sub>O<sub>3</sub>+SnO<sub>2</sub> of 0 to 13 wt%, Na<sub>2</sub>O+K<sub>2</sub>O+Li<sub>2</sub>O of 0 to 19 wt%, BaO of 0 to 26 wt%, and CaO+MgO+SrO of 0 to 13 wt%; the sixth glass powder  
10 comprises PbO of 35 to 55 wt%, B<sub>2</sub>O<sub>3</sub> of 18 to 25 wt%, ZnO of 0 to 35 wt%, BaO of 0 to 16 wt%, SiO<sub>2</sub>+Al<sub>2</sub>O<sub>3</sub>+SnO<sub>2</sub> of 0 to 9 wt%, CoO+CuO+MnO<sub>2</sub>+Fe<sub>2</sub>O<sub>3</sub> of 0 to 15 wt%, Na<sub>2</sub>O+K<sub>2</sub>O+Li<sub>2</sub>O of 0 to 19 wt%, and CaO+MgO+SrO of 0 to 13 wt%; the  
15 second filler has an average particle diameter of 0.1 to 10 µm and includes at least two oxides selected from the group consisting of NiO, Fe<sub>2</sub>O<sub>3</sub>, CrO, MnO<sub>2</sub>, CuO, Al<sub>2</sub>O<sub>3</sub>, and SiO<sub>2</sub>, which have dark colors; and the third filler  
20 has an average particle diameter of 0.1 to 10 µm and includes at least one oxide selected from the group consisting of TiO<sub>2</sub>, ZrO<sub>2</sub>, ZnO, Al<sub>2</sub>O<sub>3</sub>, BN, SiO<sub>2</sub>, and MgO, which have white colors.

When the fourth glass powder comprises ZnO of more than 48 wt%, the dielectric constant of the fourth glass powder may become too high. Further, when the fourth glass powder comprises either SiO<sub>2</sub> of more than  
25 21 wt%, or Al<sub>2</sub>O<sub>3</sub> of more than 12 wt%, or BaO+CaO+MgO+SrO of more than 15 wt%, the softening temperature of the fourth glass powder becomes so high that the fourth glass powder may be insufficiently sintered. Also, when the fourth glass powder comprises B<sub>2</sub>O<sub>3</sub> of less than 25

wt%, the softening temperature of the fourth glass powder becomes so high that the powder may be insufficiently sintered. In contrast, when the fourth glass powder comprises  $B_2O_3$  of more than 56 wt%, phase separation is apt to occur in the fourth glass powder. Further, when the fourth glass powder comprises  $Na_2O+K_2O+Li_2O$  of 0 to 38 wt%, the fourth glass powder may be crystallized.

When the fifth glass powder comprises  $PbO$  of less than 25 wt%, the fifth glass powder has such a high softening temperature that the fifth glass powder may be insufficiently sintered. In contrast, when the fifth glass powder comprises  $PbO$  of more than 65 wt%, the fifth glass powder has such a high thermal expansive coefficient that a surface of the barrier rib 140 may crack or may be bent. Further, when the fifth glass powder comprises  $ZnO$  of more than 35 wt%, the fifth glass powder has a slow viscosity change at a high temperature. When the fifth glass powder comprises either  $SiO_2$  of more than 26 wt%, or  $B_2O_3$  of more than 30 wt%, or  $Al_2O_3+SnO_2$  of more than 13 wt%, the fifth glass powder has such a high softening temperature that the fifth glass powder may be insufficiently sintered. Further, when the fifth glass powder comprises  $Na_2O+K_2O+Li_2O$  of more than 19 wt%, the fifth glass powder is apt to be crystallized. When the fifth glass powder comprises  $BaO$  of more than 26 wt%, the fifth glass powder has such a high thermal expansive coefficient, which may cause the barrier rib 140 to crack. When the

fifth glass powder comprises  $\text{CaO}+\text{MgO}+\text{SrO}$  of 0 to 13 wt%, the fifth glass powder has such a high softening temperature which may cause the fifth glass powder to be insufficiently sintered.

5           When the sixth glass powder comprises  $\text{PbO}$  of less than 35 wt%, the sixth glass powder has such a high softening temperature that the sixth glass powder may be insufficiently sintered. In contrast, when the sixth  
10   glass powder comprises  $\text{PbO}$  of more than 55 wt%, the sixth glass powder has such a high thermal expansive coefficient that a surface of the barrier rib 140 may crack or may be bent. Further, when the sixth glass powder comprises  $\text{B}_2\text{O}_3$  of less than 18 wt%, it is  
15   difficult to vitrify the barrier ribs 140. When the sixth glass powder comprises either  $\text{B}_2\text{O}_3$  of more than 25 wt%, or  $\text{BaO}$  of more than 16 wt%, or  $\text{SiO}_2+\text{Al}_2\text{O}_3+\text{SnO}_2$  of more than 9 wt%, or  $\text{CaO}+\text{MgO}+\text{SrO}$  of more than 13 wt%, the sixth glass powder such a high softening  
20   temperature which deteriorates fluidity of the sixth glass powder. When the sixth glass powder comprises either  $\text{ZnO}$  of more than 35 wt%, or  $\text{Na}_2\text{O}+\text{K}_2\text{O}+\text{Li}_2\text{O}$  of more than 19 wt%, or  $\text{CoO}+\text{CuO}+\text{MnO}_2+\text{Fe}_2\text{O}_3$  of more than 15 wt%, the sixth glass powder may be crystallized.

25           The barrier rib 140 formed of the ingredients described above has a dielectric constant of 5 to 16 and an etching rate of 2 to 50  $\mu\text{m}/\text{min}$  and can be bent at most 0.5 mm, when the barrier rib 140 has been baked for 10 to 60 minutes at 450 to 600  $^{\circ}\text{C}$ . Further, when the barrier ribs 140 have a height change of at most

1%, the barrier rib 140 has a proportion of destruction of at most 50%, which will be addressed later.

Next, a result of an experiment in which properties of the barrier rib 140 having the composition as described above have been measured will be described.

Tables 4, 5, 6 represent thermal expansive coefficients, bending, dielectric constants, and etching rates of the fourth, fifth, and sixth glass powder, which have been measured from various combinations of ingredients of the fourth, fifth, and sixth glass powder.

Table 4

Properties of the fourth glass powder measured from various combinations of ingredients of the fourth glass powder:

4 <sup>th</sup> glass powder	Proportions of ingredients (wt%)						Properties			
	ZnO	SiO <sub>2</sub>	B <sub>2</sub> O <sub>3</sub>	Al <sub>2</sub> O <sub>3</sub>	Li <sub>2</sub> O + Na <sub>2</sub> O + K <sub>2</sub> O	MgO+ CaO+ BaO+ SrO	Thermal expansive coeffic. ( $\times 10^{-7}/^{\circ}\text{C}$ )	Bend- ing	Diel- etric Const.	Etch Rate
Ex.13	30	9	25	13	8	15	60	+	13	10.2
Ex.14	0	21	27	12	40	0	112	-	15	2.0
Ex.15	30	2	56	6	1	5	74		8	4.2
Ex.16	22	21	30	5	21	1	93		17	30.9
Ex.17	40	1	46	1	12	0	68		6	16.5
Ex.18	19	6	37	0	38	0	79		20	42.0

Ex.19	48	5	25	1	6	15	73		12	21.3
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Table 5

Properties of the fifth glass powder measured from various combinations of ingredients of the fifth glass powder:

5 <sup>th</sup> glass powder	Proportions of ingredients (wt%)								Properties			
	PbO	ZnO	SiO <sub>2</sub>	B <sub>2</sub> O <sub>3</sub>	Al <sub>2</sub> O <sub>3</sub> +SnO <sub>2</sub>	Li <sub>2</sub> O +Na <sub>2</sub> O +K <sub>2</sub> O	BaO	CaO +MgO +SrO	Thermal Expan- sive coeffic. ( $\times 10^{-7}/^{\circ}\text{C}$ )	Bend- ing	Diele- ctric ratio	Etch Rate ( $\mu\text{m}/\text{min}$ )
Ex.20	25	27	26	1	13	5	0	3	72		8	3.6
Ex.21	40	22	6	10	2	20	0	1	129	-	19	48.2
Ex.22	52	14	27	1	4	1	0	1	61		11	8.2
Ex.23	26	30	4	7	8	18	0	5	103	+	15	16.8
Ex.24	41	35	15	3	2	2	0	2	78		10	18.4
Ex.25	65	6	12	8	1	6	0	2	95		11	28.2
Ex.26	27	12	1	29	1	4	26	0	65		14	21.7
Ex.27	29	31	7	17	1	7	0	8	81		13	35.1
Ex.28	41	18	8	6	7	7	0	13	100		12	38.9

Table 6

Properties of the sixth glass powder measured from various combinations of ingredients of the sixth glass powder:

6 <sup>th</sup> glass powder	Proportions of ingredients (wt%)	Properties

AMENDED SHEET (ART. 34)

	PbO	B <sub>2</sub> O <sub>3</sub>	ZnO	BaO	SiO <sub>2</sub> + Al <sub>2</sub> O <sub>3</sub> + SnO <sub>2</sub>	CoO + CuO + MnO <sub>2</sub> + Fe <sub>2</sub> O <sub>3</sub>	Na <sub>2</sub> O + CaO + SrO	MgO + CaO + SrO	Ther -mal Expan- sive coeffic. ( $\times 10^{-7}$ °/°C)	Bend -ing	Diel- ect- ric ratio	Etch Rate ( $\mu\text{m}/\text{min}$ )
Ex. 29	31	22	19	2	3	18	0	5	62	+	13	19.2
Ex. 30	55	20	0	16	1	5	1	2	96		16	42.6
Ex. 31	36	18	10	8	0	7	19	2	102		20	33.6
Ex. 32	38	22	3	14	2	7	7	7	76		12	15.4
Ex. 33	35	20	15	8	4	15	0	3	69		11	28.5
Ex. 34	46	24	1	9	1	2	2	15	112	-	15	8.4
Ex. 35	42	19	8	15	9	2	0	5	64		13	5.4
Ex. 36	36	25	33	3	0	2	0	1	102		18	21.2
Ex. 37	47	20	6	13	0	1	0	13	84		8	17.1

As apparent from Table 4, 5, and 6, the fourth, fifth, and sixth glass powder have thermal expansive coefficients, dielectric constants, and etching rates, which have values always within ranges between  $63 \times 10^{-7}$  and  $110 \times 10^{-7}/^{\circ}\text{C}$ , between 5 and 20, and between 2.0 and 50.0  $\mu\text{m}/\text{min}$ , respectively, on condition that the fourth, fifth, and sixth glass powder have ingredients mixed by the proportions described above.

Further, in order to measure the bending, paste for barrier ribs, which contains at least one glass powder selected from the group consisting of the fourth, fifth, and sixth glass powder, is applied on



the entire surface of a soda lime substrate with a size of 862 mm x 688 mm, and is then baked. When the baked paste is convexly bent at least 500  $\mu$ m, the bending is marked as "+". In contrast, when the baked paste is  
5 concavely bent at least 500  $\mu$ m, the bending is marked as "-".

Next, barrier ribs made by mixing one glass powder selected from the group consisting of the fourth, fifth, and sixth glass powder with one filler selected  
10 from the group consisting of the second and third fillers will be described.

The second filler has a function of increasing the image contrast of a PDP, but may decrease the brightness thereof. Therefore, the second filler and  
15 the third filler may be selectively used according to necessity.

When a ratio of volume of the second filler with respect to volume of glass powder for the barrier rib is at most 0.05, the mixture has a good etching  
20 uniformity but a bad optical absorption ratio which deteriorates the contrast in a driven PDP. In contrast, when the ratio is at least 0.67, the mixture has a good optical absorption ratio but a bad etching uniformity. Hereinafter, the optical absorption ratio and the  
25 etching uniformity will be described with reference to FIG. 2.

First, when  $r$  implies a width of an uppermost portion of a barrier rib,  $r'$  implies a mean value of  $r$ , and  $R$  implies a range of  $r$ , the etching uniformity is

defined as a percentage (%) calculated by a formula  
[(R/r')×100], that is, an etching uniformity (%) =  
[(R/r')×100]. Further, the optical absorption ratio is  
defined by an equation, an optical absorption ratio =  
5 {100% - (optical reflectance) - (optical  
transmissivity)}. Further, when f implies a ratio of  
volume of the second filler with respect to volume of  
glass powder for the barrier rib, the optical  
absorption ratio is defined by another equation, an  
10 optical absorption ratio = (f/0.1). Herein, when the  
etching uniformity is smaller than or equal to 7%, the  
barrier rib is usable and has a good quality.

As shown in FIG. 2, when the barrier ribs have  
been made by mixing the second filler with the fifth  
15 glass powder according to example 25 by a volumetric  
ratio of 0.05 to 0.67, the barrier ribs have an etching  
uniformity of at most 7 and an optical absorption ratio  
of at least 1. Therefore, the barrier ribs according to  
the present embodiment have a good quality.

20 Further, the etching uniformity abruptly increases  
when the volumetric ratio is larger than 0.67, while it  
decreases when the volumetric ratio is smaller than  
0.05. However, when the etching uniformity is too low,  
it is difficult to intercept colored light radiated  
25 from phosphors applied on an adjacent barrier rib, so  
that color mixing may occur.

The third filler can be classified into two kinds  
of oxides, which have weak and strong chemical  
durability with respect to acid-based etching solution,

respectively. The first kind of oxide having a weak chemical durability with respect to acid-based etching solution reacts with glass powder while it is baked, thereby deteriorating the chemical durability of the reacted glass powder. In contrast, the second kind of oxide having a strong chemical durability with respect to acid-based etching solution reacts with glass powder while it is baked, thereby increasing the chemical durability of the reacted glass powder. Further, when the ratio of volume of the third filler with respect to volume of the glass powder of the barrier ribs is smaller than 0.05, such a small proportion of the third filler decreases the white degree, to make it difficult to intercept colored light radiated from phosphors applied on an adjacent barrier rib, thereby allowing color mixing to occur. Further, when the volumetric ratio is larger than 0.67, quantity of the third filler, which does not react with the oxide, increases, so that firing strength deteriorates.

An etching rate of barrier ribs, which are made from the fourth glass powder and the third filler employing  $\text{TiO}_2$ , will be described hereinafter with reference to FIG. 3. The etching rate is defined as a total value per minute, which includes quantity of an etched portion by the etching solution, quantity of an unbaked portion separated by the ultrasonic washing, and quantity of a portion of the barrier ribs which has a reduced firing strength due to the etching.

As shown in FIG. 3, when barrier ribs having a

volumetric ratio of 0.05 to 0.67 is baked at a temperature between 450 and 600 °C, the barrier ribs can have an etching rate always between 2.0 and 50 µm/min, which is a desired range.

5 Next, measured dielectric constants, etching rates, bending, height changes, destruction ratios of barrier ribs according to types of fillers of the barrier ribs, number of the barrier ribs, and ratios of volume of the third filler with respect to volume of  
10 the glass powder of the barrier ribs will be described with reference to Table 7.

Table 7

15 Properties of barrier ribs according to types of fillers, number of the barrier ribs, and volumetric ratios of glass powder:

Class	Type of Partitions wall	Type of filler	Volumetric ratio of Filler /glass powder	Properties				
				Dielectric constant	Etch rate (µm/min)	Bending (mm)	Height difference (%)	Destruction ratio(%)
Ex.1	Singl e layer	Al <sub>2</sub> O <sub>3</sub>	0.52	10.1	28	0.21	< 0.5	9
Ex.2	Singl e layer	Spinel	0.23	12.3	17	0.24	< 0.5	21

Ex.3	Upper layer	TiO <sub>2</sub> + Al <sub>2</sub> O <sub>3</sub>	0.47	10.4	25	0.02	< 0.5	10
	Lower layer	TiO <sub>2</sub> + Al <sub>2</sub> O <sub>3</sub>	0.32	10.2				
Ex.4	Upper layer	Spinel +	0.35	15.2	22	0.13	< 0.5	15
	Lower layer	Al <sub>2</sub> O <sub>3</sub> TiO <sub>2</sub> + Al <sub>2</sub> O <sub>3</sub>	0.35	11.3				

The spinel in Table 7 implies spinel-based compound oxide.

As shown in Table 7, the dielectric constants and etching rates of the barrier ribs belong to ranges of 5 to 20 and 2.0 to 50.0  $\mu\text{m}/\text{min}$ , respectively, which implies that they have desired values.

When the bending of the glass substrate including the barrier ribs is large, it is difficult to attach the front plate and the rear plate to each other, and the PDP may be distorted even after the front plate and the rear plate are attached to each other. Herein, it is preferred that the bending of the glass substrate including the barrier ribs is at most 1 mm, the bending of the glass substrate including the barrier ribs according to the present embodiment is only 0.3 mm. Therefore, it can be said that the barrier ribs according to the present embodiment are excellent.

The height change is defined as  $\{(h1-$

h<sub>2</sub>)/h<sub>2</sub>}×100], wherein h<sub>1</sub> implies height of the barrier ribs formed through etching by means of acid-based etching solution and h<sub>2</sub> implies height of the barrier ribs measured after the barrier ribs formed through the etching are baked at 510°C for one hour. When the height change is more than 1%, it is difficult to manufacture a PDP because the barrier ribs undergo size change while the phosphorous layer is baked and the front plate and the rear plate are attached to each other after the barrier ribs are formed. Herein, all of the barrier ribs formed according to the present embodiment show a height change of at most 0.5%.

In order to examine the destruction ratio of the barrier ribs, the barrier ribs are first formed through etching by means of acid-based etching solution and are then put on a predetermined structure. Then, an iron rod, which weighs 500g and has an end portion shaped like a sphere having a radius of 3 mm, is dropped one hundred times vertically onto the uppermost surfaces of the barrier ribs from 5 mm above the uppermost surfaces. Then, the barrier ribs and the structure are inspected with an inclination of 10 to 30° by naked eye. Herein, the destruction ratio is defined as the number of barrier ribs which have been either deformed or destroyed. When the destruction ratio is larger than 50%, the barrier ribs may be destructed by vibration and impact while a completed PDP having the barrier ribs is moved or used.

Next, the phosphorous layers 150 formed on the

upper surfaces of the barrier ribs 140 will be described. The phosphorous layers 150 include red, green, and blue phosphorous layers.

5 The red phosphorous layer comprises at least two of YOx, GdOx, BOx, and EuOx, and radiates red visible rays according to an electric signal. The green phosphorous layer comprises at least one of ZnOx, SiOx, MnOx, YOx, BOx, TbOx, BaOx, and AlOx, and radiates  
10 green visible rays according to an electric signal. Further, the blue phosphorous layer comprises at least two of BaOx, MgOx, AlOx, SrOx, MnOx, and EuOx, and radiates blue visible rays according to an electric signal. Therefore, in the phosphorous layers 150, color temperatures are maintained between 8,000K and 13,000K.

15 The proportions of ingredients of the red, green, and blue phosphorous layers have degrees of freedom according to color coordinates determined by the efficiency of each phosphorous layer and the area on which each phosphorous layer is applied. Therefore,  
20 there is no restriction on the proportions of ingredients of the phosphorous layers.

Next, electrical, optical, and mechanical characteristics of a rear plate of a PDP according to the present embodiment will be compared with those of  
25 the conventional rear plate.

Table 8

Dimensions of each functional layer of a rear plate of the present invention and the conventional rear plate:

Class		Conventional rear plate	Rear plate according to the invention				
			1	2	3	4	5
Dielectric layer	Thickness ( $\mu\text{m}$ )	20	20	21	20	20	20
Barrier rib	Height ( $\mu\text{m}$ )	132	130	132	130	131	130
	Upper width ( $\mu\text{m}$ )	75	75	74	75	74	75
	Pitch ( $\mu\text{m}$ )	420	420	420	420	420	420
	Manufacturing method	Sand blasting	Etching	Etching	Etching	Etching	Etching

In the conventional rear plate shown in Table 8, dedicated materials for sandblasting have been used for the electrodes, dielectric layer, and barrier ribs. In examples 1 to 5 according to the present invention, the electrodes 120 have been made from material, which includes Ag powder of 97 wt% and glass powder of 3 wt%, the dielectric layer 130 has been made from dielectric corresponding to example 4 in Table 3, and the barrier ribs 140 have been made from the material corresponding to example 3 in Table 7.

Herein, a PDP employing a rear plate according to the present invention has the same drive circuit as that of a PDP employing the conventional rear plate. Further, a process of attaching a rear plate according to the present invention to a front plate is the same as the conventional process thereof. "Pitch" in Table 8 implies a distance between centers of adjacent two



barrier ribs.

Properties of PDPs which employ the conventional rear plate and a rear plate according to the present invention, respectively, will be described with reference to Table 9.

Table 9

Properties of PDPs employing the conventional rear plate and a rear plate according to the present invention, respectively:

Class Properties		Convent- ional Rear plate	Ex. 1	Ex. 2	Ex. 3	Ex. 4	Ex. 5
Opti- cal Proper -ty	White Peak brightness	100%	132%	135%	133%	134%	132%
	Color temper. (K)	8500	8802	8758	8773	8782	8793
	Contrast	100%	128%	134%	131%	130%	129%
Elect- ric proper -ty	Voltage Margin	100%	145%	142%	143%	143%	144%
	Power consumption	100%	89%	88%	88%	89%	90%
	Module efficiency	100%	125%	127%	127%	126%	125%
Reliab -ility	High & low temperature mis-discharge	No	No	No	No	No	No

AMENDED SHEET (ART. 34)

	Vibration & drop test	No progressiv e defect	No progr essiv e defec t	No progr essiv e defec t	No progr essiv e defec t	No progr essiv e defec t	No progr essiv e defec t
Other	Noise(dB)	100%	75%	76%	76%	77%	75%
proper ties	Panel bending.(mm)	0.7	0.3	0.2	0.1	0.2	0.2

As shown in Table 9, a PDP employing a rear plate according to the present invention shows improvements, which include about 30% in white brightness, about 300K in color temperature, about 30% in contrast, about 45% in voltage margin, and about 25% in PDP efficiency, in comparison with a PDP employing a conventional rear plate. Further, power consumption is reduced about 10%, and noise is reduced about 25%.

#### Industrial Applicability

As can be seen from the foregoing, in a rear plate of a plasma display panel according to the present invention, barrier ribs are formed through etching after backing, and thus the completed barrier ribs are not deformed. Therefore, each electrode can be exactly located on a central portion between barrier ribs.

Further, when a PDP having front and rear plates attached to each other has been completed, optical characteristics of the PDP such as white brightness,

color temperature, and contrast, and electric characteristics of the PDP such as voltage margin, power consumption, and electric efficiency, are improved, so that the reliability is improved.

5           While this invention has been described in connection with what is presently considered to be the most practical and preferred embodiment, it is to be understood that the invention is not limited to the disclosed embodiment and the drawings, but, on the  
10       contrary, it is intended to cover various modifications and variations within the spirit and scope of the appended claims.

**Claims**

1. A rear plate of a plasma display panel, the rear plate comprising:

5 a glass substrate;  
electrodes formed in a shape of patterns on an upper surface of the glass substrate;

a dielectric layer formed on upper surfaces of the electrode and the upper surface of the glass substrate;

10 barrier ribs formed in a shape of a pattern through etching on an upper surface of the dielectric layer; and

phosphorous layers formed on side surfaces and bottom surfaces of the barrier ribs and including red,  
15 green, and blue phosphorous layers, which emit red, green, and blue light according to an electric signal, respectively, wherein:

the electrodes are made from mixture of a conductive metal powder of 51 to 99.5 wt% and a first  
20 glass powder of 0.5 to 49 wt%, the conductive metal powder being at least one kind of metal powder selected from metal powder of An, Ag, Pt, Pd, Ni, and Cu, the conductive metal powder having an average particle diameter of 0.1 to 7  $\mu\text{m}$ , the first glass powder having  
25 an average particle diameter of 0.5 to 10  $\mu\text{m}$  and a specific resistance of  $1.0 \times 10^{-6}$  to  $5.0 \times 10^{-6} \Omega\text{cm}$ ;

the dielectric layer is made from mixture of a first filler and at least one glass powder selected from among a second glass powder and a third glass

powder, the second glass powder including PbO of 30 to 80 wt%, ZnO of 0 to 20 wt%, SiO<sub>2</sub> of 0 to 20 wt%, B<sub>2</sub>O<sub>3</sub> of 5 to 40 wt%, Al<sub>2</sub>O<sub>3</sub> of 0 to 12 wt%, Na<sub>2</sub>O+K<sub>2</sub>O+Li<sub>2</sub>O of 0 to 5 wt%, and BaO+CaO+MgO+SrO of 0 to 5 wt%, the third glass powder including Bi<sub>2</sub>O<sub>3</sub> of 36 to 84 wt%, B<sub>2</sub>O<sub>3</sub> of 5 to 28 wt%, PbO of 0 to 46 wt%, ZnO of 0 to 30 wt%, Al<sub>2</sub>O<sub>3</sub> of 0 to 13 wt%, SiO<sub>2</sub> of 0 to 10 wt%, Na<sub>2</sub>O+K<sub>2</sub>O+Li<sub>2</sub>O of 0 to 5 wt%, and BaO+CaO+MgO+SrO of 0 to 3 wt%, each of the second and third glass powders having an average particle diameter of 0.5 to 10  $\mu$ m, a softening temperature of 390 to 550 °C, a thermal expansive coefficient of  $63 \times 10^{-7}$  to  $83 \times 10^{-7}/^{\circ}\text{C}$ , a dielectric constant of 11 to 26, and an etching rate of 0.1 to 1.0  $\mu\text{m}/\text{min}$ , the first filler having an average particle diameter of 0.5 to 10  $\mu\text{m}$  and including at least one oxide selected from the group consisting of TiO<sub>2</sub>, ZrO<sub>2</sub>, ZnO, Al<sub>2</sub>O<sub>3</sub>, BN, SiO<sub>2</sub>, and MgO, which are white oxides, a ratio of volume of the first filler with respect to volume of the glass powder in the dielectric layer being 0.05 to 0.30, thereby the dielectric layer having a dielectric constant of 11 to 26, a reflectance of 50 to 80%, an etching rate of 0.1 to 1.0  $\mu\text{m}/\text{min}$ , and a porosity of 5, when the dielectric layer has been baked for 10 to 60 minutes at 450 to 600 °C;

the barrier ribs are made from mixture which includes at least one glass powder selected from the group consisting of a fourth, fifth, and sixth glass powders and at least one filler selected from the group consisting of a second filler and a third filler, the

fourth glass powder including ZnO of 0 to 48 wt%, SiO<sub>2</sub> of 0 to 21 wt%, B<sub>2</sub>O<sub>3</sub> of 25 to 56 wt%, Al<sub>2</sub>O<sub>3</sub> of 0 to 12 wt%, Na<sub>2</sub>O+K<sub>2</sub>O+Li<sub>2</sub>O of 0 to 38 wt%, and BaO+CaO+MgO+SrO of 0 to 15 wt%, the fifth glass powder including PbO of 25 to 65 wt%, ZnO of 0 to 35 wt%, SiO<sub>2</sub> of 0 to 26 wt%, B<sub>2</sub>O<sub>3</sub> of 5 to 30 wt%, Al<sub>2</sub>O<sub>3</sub>+SnO<sub>2</sub> of 0 to 13 wt%, Na<sub>2</sub>O+K<sub>2</sub>O+Li<sub>2</sub>O of 0 to 19 wt%, BaO of 0 to 26 wt%, and CaO+MgO+SrO of 0 to 13 wt%, the sixth glass powder including PbO of 35 to 55 wt%, B<sub>2</sub>O<sub>3</sub> of 18 to 25 wt%, ZnO of 0 to 35 wt%, BaO of 0 to 16 wt%, SiO<sub>2</sub>+Al<sub>2</sub>O<sub>3</sub>+SnO<sub>2</sub> of 0 to 9 wt%, CoO+CuO+MnO<sub>2</sub>+Fe<sub>2</sub>O<sub>3</sub> of 0 to 15 wt%, Na<sub>2</sub>O+K<sub>2</sub>O+Li<sub>2</sub>O of 0 to 19 wt%, and CaO+MgO+SrO of 0 to 13 wt%, each of the fourth, fifth, and sixth glass powders having an average particle diameter of 0.5 to 10 μm, a softening temperature of 390 to 630 °C, a thermal expansive coefficient of  $63 \times 10^{-7}$  to  $83 \times 10^{-7}/^{\circ}\text{C}$ , a dielectric constant of 5 to 20, and an etching rate of 2.0 to 50.0 μm/min, the second filler including at least two oxides selected from the group consisting of NiO, Fe<sub>2</sub>O<sub>3</sub>, CrO, MnO<sub>2</sub>, CuO, Al<sub>2</sub>O<sub>3</sub>, and SiO<sub>2</sub>, which have dark colors, the third filler including at least one oxide selected from the group consisting of TiO<sub>2</sub>, ZrO<sub>2</sub>, ZnO, Al<sub>2</sub>O<sub>3</sub>, BN, SiO<sub>2</sub>, and MgO, which have white colors, each of the second and third fillers having an average particle diameter of 0.1 to 10 μm, a ratio of the volume of the filler with respect to the volume of the glass powder for the barrier ribs being 0.05 to 0.67, thereby the barrier ribs having a dielectric constant of 5 to 16 and an etching rate of 2 to 50 μm/min and enabling the glass

substrate having the barrier ribs to have a bending of  
at most 0.3 mm, when the barrier ribs have been baked  
for 10 to 60 minutes at 450 to 600 °C, the barrier ribs  
having a height difference of at most 1% when the  
5 barrier ribs has been baked at 510°C for one hour after  
being etched by acid-based etching solution, the  
barrier ribs having a destruction ratio of 50% when an  
iron rod, which weighs 500g and has an end portion  
shaped like a sphere having a radius of 3 mm, is  
10 dropped one hundred times vertically onto uppermost  
surfaces of the barrier ribs from 5 mm above the  
uppermost surfaces, each of the barrier ribs having at  
least one layer; and

the red phosphorous layer includes at least two of  
15 YOx, GdOx, BOx, and EuOx, the green phosphorous layer  
includes at least one of ZnOx, SiOx, MnOx, YOx, BOx,  
TbOx, BaOx, and AlOx, and the blue phosphorous layer  
includes at least two of BaOx, MgOx, AlOx, SrOx, MnOx,  
and EuOx, so that, in the phosphorous layers, color  
20 temperatures are maintained between 8,000K and 13,000K.